# **Vidyasagar University**

# Curriculum for Industrial Chemistry (Major) [Choice Based Credit System]

#### Semester-I

Course	Course Code	Name of the Subjects	Course Type/ Nature	Teaching Scheme in hour per week			Credit	Marks
				L	T	P		
CC1		C1T: Unit Process for Organic Synthesis and Industrial Applications	Core Course-1	5	1	0	6	75
CC2		C2T: Inorganic Materials for Chemical Industries and Industrial Waste Management	Core Course-2	5	1	0	6	75
GE1		TBD	Generic Elective-1				4/5 2/1	75
AECC		English	AECC (Elective)	1	1	0	2	50
Semester Total							20	275

L=Lecture, T=Tutorial, P=Practical, CC- Core Course, TBD - To be decided, AECC-Ability Enhancement Compulsory Course

**Generic Elective** (**GE**) (Interdisciplinary) from other Department [Paper will be of 6 credits]. Papers are to be taken from following discipline: **Computer Science/Mathematics/Physics/Chemistry/Economics** 

Modalities of selection of Generic Electives (GE): A student shall have to choose 04 Generic Elective (GE1 to GE4) strictly from 02 subjects / disciplines of choice taking exactly 02 courses from each subjects of disciplines. Such a student shall have to study the curriculum of Generic Elective (GE) of a subject or discipline specified for the relevant semester.

# Semester –I Core Course (CC)

# CC-1 : Unit Process for Organic Synthesis and Industrial Applications

Credits 06

# C1T: Unit Process for Organic Synthesis and Industrial Applications

#### **Course Contents:**

#### Nomenclature Generic name, trade name. Raw Material Resources:

Cellulose, Starch Properties, Modification, Important Industrial Chemical derived from them, alcohol and alcoholbased chemical, Oxalic acid,

# **Unit Process in Organic Chemical Manufacture Nitration:**

Introduction – Nitrating agents, Kinetic and mechanism of nitration process such as nitration of –

- a) Paraffinic hydrocarbons
- b) Benzene to Nitrobenzene and m-dinitrobenzene
- c) Chlorobenzene to O-& P-nitrobenzene
- d) Acetanilide to P-nitroacetanilide
- e) Toluene, Continuousvs. batch nitration

### **Halogenations:**

Introduction-kinetics of halogenations reactions. Reagents for halogenations, alogenations of aromatics-side chain and nuclear halogenations, Commercial Manufactures-chlorobenzenes, chloral, monochloractic and chloromethane, dichloroflouromethane.

# **Sulphonation:**

Introduction-sulophonating agents, Chemical and physical factors in sulphonation, Kinetics and mechanism of sulphonation reaction, commercial sulfonation of benzene, naphthalene, alkyl benzene, Batch vs.continuous sulfonation.

# **Unit Processing in Organic Synthesis**

# **Oxidation:**

Introduction-Types of oxidation reactions, Oxidizing agents, Kinetics and mechanism of oxidation of organic compounds, Liquid phaseoxidation, vapour phase oxidation, Commercial manufacture of benzoicacid maleic anhydride, phthalic anhydride, acrolein, acetaldehyde, aceticacid.

#### **Hydrogenation:**

Introduction-Kinetics and thermodynamics of hydrogenation reaction, Catalysts of hydrogenation reaction, Hydrogenation of vegetable oil, Manufacture of methanol from carbon monoxide and hydrogen, hydrogenation of acids to alcohols, catalytic reforming.

#### **Alkylation:**

Introduction, Types of alkylation, alklating agents,thermodynamic and mechanism of alkylation reactions, Manufacture of alkyl benzenes (for detergent manufacture), ethyl benzene, phenyl ethylalcohol, (N-alkyl anilines mono and di-methyl and ethyl anilines).

**Esterification:** Introduction, Hydrodynamics and kinetics of esterificationreaction, terification by organic acids, by addition of unsaturated compounds, esterification of carboxy acid derivatives, commercial manufacture of ethyl acetate, dioctyl phthalate, vinylacetate, cellulose acetate.

#### **Amination:**

- a) By Reduction: Introduction, Method of reduction-metal and acid, catalytic, sulfide, lectrolytic, metal and alkali sulfites, metalhydrides, sodium metal and alkali sulfites, metal hydrides, sodiummetal, concentrate caustic oxidation, reduction, commercialmanufacture of aniline, m-nitro aniline, p-amino phenol.
- b) By Aminolysis: Introduction, Animating agents, Kinetics,thermodynamics and mechanism of hydrolysis.

# CC-2: Inorganic Materials for Chemical Industries and Industrial Waste Management Credits 06

C2T: Inorganic Materials for Chemical Industries and Industrial Waste Management

#### **Course Contents:**

#### **Material Science:**

Mechanical properties of materials and change with respect to temperature. Materials of constructions used in industry.

#### **Metals and alloys:**

Important metals and alloys, Iron, Copper, Aluminum, Lead, Nickel, Titanium and their alloys phen diagram Mechanical and chemical properties and their applications.

#### **Cement:**

Types of cement, composition, manufacturing process setting of

#### **Ceramics:**

Introduction, Types, manufacturing process, applications, refractoriness, concept of bio ceramics.

# **Polymeric Materials:**

Commodity polymers, blends and composites their constitution, chemical and physical properties, industrial applications.

#### Glass:

Types, composition, manufacture, physical and chemical properties, Industrial pplications.

**Corrosion:** Various types of corrosion relevant to chemical industry mechanism, reventive methods.

# **Effluent Treatment and Waste Management**

Principles and equipments for aerobic, anaerobic treatment, absorption, filtration, sedimentation. Bag filters, electrostatic precipitator, mist eliminators, wet scrubbers. Absorbers . Solid waste Management. Industrial Safety Laws.

# **Industrial Aspects of Inorganic Chemistry**

# **Basic Metallurigical Operations:**

Pulverization, Calcinations, Roasting, Refining. Physicochemical principles of extraction of Iron, Copper, Lead, Silver, Sodium, Aluminum, Magnesium, Zinc, Chromium. Inorganic Materials of Industrial Importance: Their availability, orms, structure and modification, Alumuna, Silica, Silicates, Clays, Mica, Carbon, Zeolites.